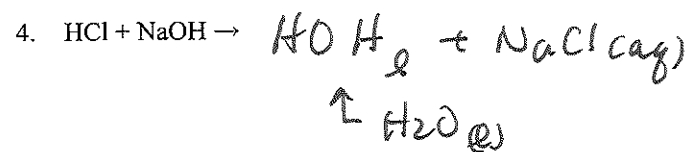
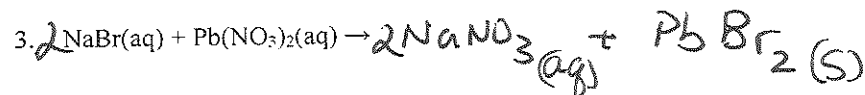
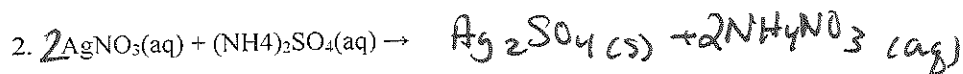
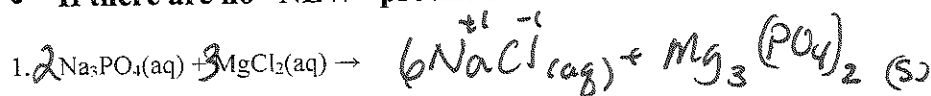


Class  
Chemistry  
Solubility Reactions #1

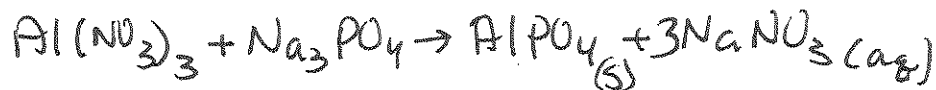
Molecular Equation:  $AB + CD \rightarrow CB_{(aq)} + AD_{(s)}$

Insoluble = precipitate Soluble = aqueous	Quick Check $Na^+X$ $K^+X$ $NH_4^+X$ $XNO_3^-$	Always Soluble
***Need at least 1 insoluble product or water		

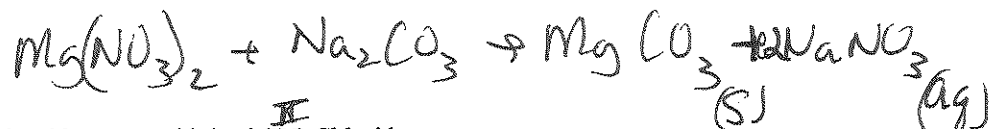
- Determine potential products
- Indicate if the product will be soluble(s), insoluble (aq), or Liquid (H<sub>2</sub>O(l))
- If there are no "NEW" products indicate no reaction.



5. Aluminum nitrate reacts with sodium phosphate.



6. Magnesium Nitrate reacts with Sodium Carbonate



7. Potassium Hydroxide reacts with lead (II) Chloride

